Cloning and Expression of the Gene Encoding Flavodoxin from Desulfovibrio vulgaris (Miyazaki F)¹

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The gene encoding a flavodoxin of *Desulfovibrio vulgaris* (Miyazaki F) was cloned, and overexpressed in *Escherichia coli*. A 1.6-kbp DNA fragment, isolated from *D. vulgaris* (Miyazaki F) by double digestion with *Sal*I and *Eco*RI, contained the flavodoxin gene and its regulatory region. An expression system for the flavodoxin gene under control of the T7 promoter was constructed in *E. coli*. The purified protein was soluble and exhibited a characteristic visible absorption spectrum. HPLC analysis of the recombinant flavodoxin revealed the presence of an identical FMN to that found in the native *D. vulgaris* flavodoxin, and its dissociation constant with FMN was determined to be 0.38 nM. *In vitro* H₂ reduction analysis indicated that the recombinant flavodoxin is active, and its redox potential was determined to be $E_1 = -434$ and $E_2 = -151$ mV using methyl viologen and 2-hydroxy-1,4-naphthoquinone, respectively. Its redox behavior was also examined with the recombinant flavodoxin adsorbed onto a graphite electrode. The mutant, A16E, was also produced, which revealed the feature of a conserved Glu residue at the surface of the molecule.

Key words: *Desulfovibrio vulgaris*, expression, flavodoxin, FMN-binding, site-directed mutagenesis.

Flavodoxins, a group of small flavoproteins of molecular masses of 15-23 kDa, have been isolated from a variety of microorganisms, in which they are thought to function as electron-transfer proteins in various metabolic pathways (1). They contain a single molecule of non covalent-bound FMN as their prosthetic group, and they exhibit a very negative reduction potential for the semiquinone/hydro-quinone couple of the bound FMN cofactor. This negative reduction places these proteins in the potential range of ferredoxins, for which they can substitute *in vitro* and *in vivo* (2, 3). The redox properties of bound FMN in flavo-doxins differ from those of protein-free FMN, in that the semiquinone state is stable.

Three-dimensional structures have been determined for

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the flavodoxins from Clostridium beijerinckii MP (4) and Chondrus crispus (5) by X-ray crystallography, and for the Megasphaera elsdenii flavodoxin by two-dimensional NMR spectroscopy (6). The structure of the flavodoxin from Desulfovibrio vulgaris (Hildenborough) has been well studied in its three redox states by X-ray crystallography or NMR spectroscopy (7-10). The data have shown a multitude of interactions between the flavin and the apoprotein, and have suggested mechanisms by which the apoprotein might modulate the redox properties of the flavin.

Sulfate-reducing bacteria contain many redox proteins, and flavodoxins have been isolated from six strains and their amino acid sequences have been determined (11-15). Much is known about many redox proteins of *D. vulgaris* (Miyazaki F) (16, 17), but little has been reported on the flavodoxin from this strain, and even its primary structure has not been determined yet (18). On the other hand, another flavoprotein, an FMN-binding protein, has been isolated only from this strain, and it has been found to consist of 122 amino acids (19). The semiquinone state of the FMN-binding protein is unstable, and the redox potential between the oxidized state and the fully reduced state has been estimated to be -325 mV. So, the FMN-binding protein appears to be different from other flavodoxins from

¹ The nucleotide sequence data reported in this paper will appear in the DDBJ, EMBL, and GenBank nucleotide sequence databases under accession number D88439.

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Abbreviations: ORF, open reading frame; PAGE, polyacrylamide gel electrophoresis; UV/VIS, ultraviolet/visible; BPG, basal-plane pyrolytic graphite; EPG, edge-plane pyrolytic graphite; DMPC, L- α -phosphatidylcholine dimyristoyl; CV, cyclic voltammogram; DPV, differential-pulse voltammogram.

sulfate-reducing bacteria in terms of redox properties, but it is still unknown whether or not the FMN-binding protein is the flavodoxin of *D. vulgaris* (Miyazaki F).

We have also carried out a series of genetic studies on the redox proteins of D. vulgaris (Miyazaki F) and the previously cloned cytochrome c_3 gene (20), the FMN-binding protein gene (19), the cytochrome c-553 and cytochrome c oxidase-like protein genes (21), and the rubredoxin and desulfoferrodoxin genes (22). In the present paper, we report the molecular cloning and sequencing of the gene encoding the flavodoxin of D. vulgaris (Miyazaki F), its overproduction in Escherichia coli, and some of its characteristics as a holoprotein. A preliminary account of this work was presented previously (23).

EXPERIMENTAL PROCEDURES

Bacterial Strains and Materials—E. coli strain JM109, recA 1, $\Delta(lac\text{-}proAB)$, end A 1, gyrA96, thi-1, hadR 17, relA 1, supE 44[F'traD36, proAB⁺, lacI^q Z Δ M15], was used for cloning, and strain BL21(DE3), F⁻, ompT [lon]hsdS_B (rB⁻mB⁻; an E. coli B strain) with DE3, a λ prophage carrying the T7 RNA polymerase gene, was used for expression of the flavodoxin gene. D. vulgaris (Miyazaki F) was grown (24) and used for DNA isolation. Restriction endonucleases and DNA-modifying enzymes were purchased from Nippon Gene and Takara Shuzo, respectively. [γ -³²P]ATP (185 TBq/mmol) was obtained from ICN. All other chemicals were of analytical grade for biochemical use.

Cloning and Sequencing-Genomic DNA isolated from D. vulgaris (Miyazaki F) was prepared by the method of Saito and Miura (25). We performed PCR using genomic DNA as the template. Two primers were designed for this reaction according to the conserved regions of the published amino acid sequences of flavodoxins from other sulfatereducing bacteria because the amino acid sequence of that from D. vulgaris (Miyazaki F) was unknown (see Fig. 3). The nucleotide sequences were as follows: primer-1 was a 23 mer, and its nucleotide sequence was 5'-GGCTCCACC-ACCGGCAACACGGA-3', corresponding to the amino acid sequence of °G-S-T-T-G-N-T-E. Primer-2 was a 20 mer, and its nucleotide sequence was 5'-GTCCACCGCGCCGC-AGAAGT-3', which is the complementary sequence of the corresponding amino acid sequence of ¹⁰⁰Y-F-C-G-A-V-D. These sequences were based on the codon usage of previously known genes in the same strain. The PCR mixture was phosphorylated with T4 polynucleotide kinase and then ligated with pUC18 previously cut with SmaI. We extracted the plasmids from several E. coli JM109 cells transformed with this reaction mixture and determined the nucleotide sequence of the inserted DNA. One of the nucleotide-derived amino acid sequences was similar to the amino acid sequence of another flavodoxin from sulfate-reducing bacteria. We then synthesized a 30-mer probe DNA, with the sequence of 5'-GTGTAGCTGCTGTCGCCGCAG-CCGAAGCAG-3', which is the complementary sequence of the corresponding the amino acid sequence of ⁸⁹A-C-F-G-C-G-D-S-S-Y-T. We carried out Southern hybridization with this labeled oligonucleotide at 65°C and detected a band hybridizing to a ca. 1.6-kilobase pair Sall-EcoRI fragment by use of a BAS1000, FUJIX (data not shown). So we digested the genomic DNA with Sall and EcoRI, and the

digest was separated into several fractions on an agarose gel according to size. The separated fragments were ligated into the SalI-EcoRI site of pUC18, and E. coli JM109 was transformed with the resulting ligation mixture. One such transformant was found by means of the colony hybridization method to harbor a plasmid carrying the ca. 1.6-kbp SalI-EcoRI fragment of D. vulgaris (Miyazaki F) DNA, which was named pSE1600. The nucleotide sequence of the inserted fragment was determined by sequencing of its restriction fragments which were cloned into the multicloning site of pUC18, and a deletion mutant was obtained by use of exonuclease III and Mung Bean nuclease. We used the dideoxy chain termination method (26), for which we used a DNA sequencer (A.L.F. II; Pharmacia LKB).

High-Level Expression of Flavodoxin—We attempted to construct a high level expression system in *E. coli* using pUT7. Expression vector pUT7 has a T7 promoter and a high copy number (27). Plasmid pUT7 was digested with *Bam*HI, and then blunt-ended and digested with *NcoI*. Another plasmid, pSE1600, was digested with *Eco*RI and blunt-ended and then cut with *NcoI*, because *NcoI* consists of a region of the initiation codon of the flavodoxin gene and *Eco*RI downstream of the gene (see Fig. 1). The resultant plasmid was designated as pUTFLV.

Site-Directed Mutagenesis—Position 16 of this flavodoxin is Ala, whereas Glu is conserved at position 16 in all other flavodoxins from sulfate-reducing bacteria. It is located near the FMN-binding region, so it is thought to be important in the binding of FMN. Site-directed mutagenesis was carried out using the modified method of Inouye (28), and we used an 18-mer deoxyoligonucleotide, of which the nucleotide sequence was 5'-CAACACCGAATGGGTCGC-3'. We obtained mutant plasmid pSE1700, and its insertion was ligated to pUT7 by the same procedure as used for pUTFLV. Finally, mutant plasmid pUTFLV2 was obtained, which encoded the A16E mutant flavodoxin.

Purification of the Expressed Protein-E. coli BL21-(DE3) cells transformed with pUTFLV were cultured for 9 h at 37°C in 1.7 ml LB containing 50 μ g/ml ampicillin. Eight flasks containing 250 ml of this medium were incubated with 1.7 ml of the culture overnight with shaking at 37°C. Cells were harvested by centrifugation at 6,000 rpm for 15 min. The cell pellet was suspended in 10 mM Tris-HCl (pH 8.0) and sonicated using a Model 201M sonicator (KUBOTA) at 9,000 Hz and 200 W for 10 min. After the suspension had been centrifuged at 35,000 rpm for 2 h at 4°C, the gray supernatant was dialyzed against distilled water overnight at 4°C, and the resulting yellow solution was loaded onto a DEAE-cellulose (DE52) column (2.0 \times 9.0 cm) equilibrated with 10 mM Tris-HCl (pH 8.0). The column was washed with 100 ml of the buffer and then developed with a gradient of 0 to 300 mM NaCl in 10 mM Tris-HCl (pH 8.0). Flavodoxin-containing fractions were identified from their optical spectra. The colored fractions were collected and diluted twice, and then loaded onto the DE52 column again under the same conditions. The eluted colored solution $(A_{268}; A_{448} < 4.5)$ was dialyzed against distilled water and then lyophilized. Gel-filtration on a Superose HR-12 column (1.0×30.0 cm) was carried out using a Pharmacia FPLC system. The eluent was 200 mM NaCl/10 mM Tris-HCl (pH 8.0) and the flow rate was 0.4 ml/min. The purified recombinant flavodoxin was then eluted for 35 min. SDS-PAGE was carried out according to

the method of Laemmli (29) with a gel concentration of 15%.

Specroscopic Analysis and N-Terminal Amino Acid Sequencing—Absorption spectra were recorded at room temperature with a Hitachi 320 spectrophotometer. For the amino acid sequence analysis, we carried out an additional separation. The protein purified by gel filtration was dissolved in distilled water, and then subjected to reverse phase HPLC (RESOURCE-RPC: Pharmacia, 3 ml) with a linear gradient of acetonitrile in 1% TFA at the flow rate of 2 ml/min using a Shimadzu HPLC system. The purified peptide was then analyzed using an Applied Biosystems model 476A protein sequenator.

Determination of Oxidation/Reduction Potentials and In Vitro Activity—The redox potentials for the flavodoxin semiquinone/fully reduced flavodoxin (E_1) and oxidized flavodoxin/flavodoxin semiquinone (E_2) couple were determined by means of equilibrium reactions and spectrophotometric measurements (30) with mixtures of flavodoxin, methyl viologen and 2-hydroxy-1,4-naphthoquinone, respectively. The redox potential, E_n , for the system at equilibrium was calculated with the Nernst equation,

$$E_{\rm n} = E_{\rm m,7}(\rm dye) + \frac{RT}{nF} \ln \frac{[\rm oxidized dye]}{[\rm reduced dye]}$$

where R is the gas constant, T the absolute temperature, Fthe Faraday, and *n* the number of electrochemical equivalents. A solution of flavodoxin and a dye in 50 mM sodium phosphate buffer, pH 7, and 3 mM EDTA in a closed allglass cuvette was made anaerobic by repeated cycles of evacuation and flushing with oxygen-free nitrogen. For the determination of E_1 , we monitored the absorbance at 600 and 730 nm. Methyl viologen in the oxidized state absorbs at neither 600 nor 730 nm, where flavodoxin in the fully reduced state, and the semiquinone state does not absorb at 730 nm. ε_{600} of flavodoxin in the semiquinone state and the fully reduced state used was at 4,870 and 900 $M^{-1} \cdot cm^{-1}$, respectively. For the determination of E_2 , preliminary experiments showed an isosbestic point between the oxidized and semiquinone states at 506 nm ($\varepsilon_{506} = 3,160 \text{ M}^{-1}$. cm⁻¹). Dye reduction was measured at 620 nm ($\varepsilon_{620} = 4,020$ $M^{-1} \cdot cm^{-1}$), where neither the oxidized flavodoxin nor the dye absorb. For the determination of E_1 , an amount of a dithionite solution was added to the dye along with the flavodoxin in the semiquinone state, all of which were photo-reduced using a 150 W slide projector (31). For E_2 , the dye and oxidized flavodoxin solution mixture was photo-reduced by the same procedure as for E_1 , and the reduction of the dye and the flavodoxin was determined after each period of photo irradiation.

The reduction for the H₂ assay was carried out under anaerobic conditions. The reaction mixture, with a total volume of 1 ml, contained hydrogenase, cytochrome c_3 and 20 μ M flavodoxin in 30 mM phosphate buffer (pH 7.0).

Redox Behavior on Bare and Phospholipid-Modified Graphite Electrodes—We measured the redox behavior on a bare basal-plane pyrolytic graphite (BPG) or an edgeplane pyrolytic graphite (EPG) electrode. Voltammetric measurements were conducted using a cell with a threeelectrode configuration under a nitrogen atmosphere at $22\pm 2^{\circ}$ C in 30 mM phosphate buffer (pH 7.3). An Ag/AgCl electrode in a saturated KCl solution was used as the reference electrode, although all of the potentials cited in the present paper were referenced to a normal hydrogen electrode. The procedure was that described in detail previously (19) with slight modification. As demonstrated in our previous paper (22), rubredoxin exhibited a fair redox response on a phospholipid-modified graphite electrode. It is expected that a phospholipid film provides a suitable reaction environment for electrode reactions of electron transfer proteins, probably due to the ability of the film to prevent structural changes of the proteins. Thus, the electrochemistry of flavodoxin on a phospholipid-modified BPG electrode was examined. To modify the BPG electrode surface with a phospholipid, $L \cdot \alpha$ -phosphatidylcholine dimyristoyl (DMPC) was used. The procedures will be described in detail in a future paper.

Identification of the Prosthetic Group, and Determination of Dissociation Constants for the Complexes of Apoflavodoxin with FMN and Riboflavin—The peptide portion was removed by resolving the purified protein in ice-cold 5% trichloroacetic acid and 0.3 mM EDTA in 0.1 M phosphate buffer (pH 7.0), and centrifugation at 12,000 rpm and 4°C for 10 min. The solution containing a prosthetic group was then neutralized using diethylether and sodium bicarbonate. To identify the prosthetic group, this solution was loaded onto an HPLC C-8 column (Hiber LiChrosorb RP-8. Cica-MERCK). The eluent was 10% acetonitrile in 0.1% trifluoroacetic acid, and the flow rate was 2 ml/min. This purified FMN solution was also used for the determination of dissociation constants. The dissociation constants (K_d) for the binding of FMN and riboflavin to apoflavodoxin were determined by fluorescence titration (32). Microliter quantities of apoflavodoxin were added to 3 ml aliquots of a 0.1 μ M flavin solution in 50 mM potassium phosphate (pH 7.0) and 5 mM EDTA at 25°C. Quenching of the flavin fluorescence occurred within several minutes. The proportion of bound flavin was determined at equilibrium by measuring the extent of quenching.

Immunoblot Analysis—To compare the higher structure of the recombinant flavodoxin to that of the FMN-binding protein from the same strain, immunoblot analysis was carried out. Samples were loaded onto a 15% polyacrylamide SDS minislab gel. Blotting onto a nitrocellulose filter was performed electrophoretically, followed by blocking by shaking in 3% (w/v) bovine serum albumin in buffer A containing 10 mM Tris-HCl (pH 8.0), 150 mM NaCl, and 0.5% Tween 20 for 30 min. After a wash with buffer A, the blot was incubated with a 1:1,000 dilution of polyclonal antiserum, raised against the holo-FMN-binding protein purified from E. coli, in buffer A for 30 min (19). The blot was then thoroughly washed with buffer A and incubated with 1 μ l of alkaline phosphatase-conjugated goat antirabbit IgG in buffer A. Following two washes, first with buffer A and then with alkaline phosphatase buffer [100 mM Tris-HCl (pH 9.5), 100 mM NaCl, and 5 mM MgCl₂], the nitrocellulose filter was incubated with nitroblue tetrazolium and 5-bromo-4-chloro-3-indoyl phosphate in 5 ml of alkaline phosphatase buffer. The blot was finally washed with water and dried.

RESULTS

Cloning and Nucleotide Sequencing of the Flavodoxin Gene—The partial restriction map and sequencing strategy used for the determination of the nucleotide sequence of SalI-EcoRI are shown in Fig. 1, and the determined nucleotide sequence containing the entire flavodoxin gene is shown in Fig. 2. The open reading frame (ORF) encoding flavodoxin is composed 148 amino acids, including an amino-terminal Met, and is preceded by putative ribosome-binding site (AGGA), which has been observed to comprise nucleotides 404-407 of the SalI-EcoRI fragment. The amino acid sequence of the flavodoxin from D. vulgaris (Miyazaki F) is homologous to those from other sulfate reducing bacteria, especially that from D. vulgaris (Hildenborough), the identity being 67%. However, position 16, which is near the FMN-binding region and located on the surface of the molecule, is Ala, while position 16 in all other flavodoxins from sulfate-reducing bacteria is occupied by Glu (Fig. 3). No other protein gene was found in this cloned fragment.

Purification of the Recombinant Flavodoxin—We purified the recombinant flavodoxin from a cell lysate of *E. coli* BL21(DE3) harboring pUTFLV. Through chromatographic steps on DE52 and Superose HR12, a large amount (25 mg/ g, wet cell) of flavodoxin was purified to homogeneity on SDS-PAGE. The molecular weight in the denatured state was determined to be approximately 19,000 by SDS-PAGE (Fig. 4), which was rather different from the calculated value (15,600) based on the amino acid sequence deduced from the nucleotide sequence. The result of amino acid sequencing of the amino terminus was Ala-Asn-Val-Leu-Ile-Val-Tyr-Gly-Ser-Thr, and Met was not detected. Therefore, the molecular mass of the recombinant flavo-doxin appears to be 15,900, which includes one molecule of FMN.

Identification of the Prosthetic Group—To identify the prosthetic group, bound to the recombinant flavodoxin, the UV-visible spectrum of the purified holoprotein was measured, as shown in Fig. 5. In the visible region, absorption maxima were observed at 377 and 448 nm, which is characteristic of proteins that bind to flavin derivatives. The prosthetic group extracted with trichloroacetic acid was subjected to reverse phase HPLC on a C-8 column, and the retention time of the obtained prosthetic group was compared with those of the flavin derivatives. As shown in Fig. 6, the retention time of the prosthetic group released from the recombinant flavodoxin was exactly the same as that of FMN. The $A_{448}:A_{268}$ ratio of the holoprotein was

1000

800

400

200

Smal Sphl Ncol

600

Pstl

Flavodoxin gene

1400

1613 (bp)

Eco Bl

1200

Fig. 1. Partial restriction map and sequencing strategy for the cloned 1.6-kbp SalI-EcoRI fragment. The open reading frame is shown within the box. The arrows indicate the region sequenced on each DNA strand for the restriction fragment cloned into the multicloning site of pUC18, and the deletion mutants obtained on treatment with exonuclease III and mung bean nuclease. Some restriction enzyme sites are also indicated.

90	100	110 12	0 130	140 150	160
CTTTTTTCTTGAAAA	GGATTTTCATTT	TCAT <u>TACGGT</u> TC	CCTCAACGATCAACCTC	CAACGACGCCCCCCCAT	GCCCACA
-35 region	n	-10 regior	1		
170	180 2	190 20	0 210	220 230	240
TCACTCCCACGTATT	CCAGCCTCCGGC	CTTTCCCGTCCG	TCCCGGCCTTGCAACCO	CGGGGCGGACGGGGAACG	SCCACCCC
250	260	270 28	0 290	300 310	
CTGCACGCATGCGGC	ACGTTCCGCAGG	TTGCGCGCATCO	CCCCCTGCTCACCCTGC	CTATTGGGCCCATTGGG	TCCACCC
C100110000111000000					
330	340	350 36	0 370	380 390	400
ACCAGGCGGAACGGT	GTAGTCGGGCCG	CCTGACGGGTGG	CCCCACTCGAACACCGA	ACACGCTTCACAGACACC	CATCCACC
410	420	430 44	0 450	460 470	480
CGG <u>AGGA</u> TTCCCATG	GCCAACGTGCTC	ATCGTCTACGG	TCCACCACCGGCAACAC	CGCCTGGGTCGCCGAAA	ACCGTCGG
SD sequence M i	ANVL	IVYG	STTGNT	AWVAET	r V G
400	500	610 5 2	5 20	540 550	560
490 MCCCCACAMCCCCCA	SUU AGCGGGGCCACAG	210 CGTTGAAATCC(CGACGCGGGCCAGGTAG	000000000000000000000000000000000000000	GAAGGCC
P D T A F	A G H S	VETR		AEGLC	EGR
K D I K L					
570	580 5	590 60	610	620 630	640
GCGACCTCGTGCTGT	TCGGCTGCTCCA	CCTGGGGGCGAC	ACGAAATCGAACTGCAG	GACGACTTCATCCACCT	GTACGAA
DLVLF	GCST	WGDI	EIELQ	DDFIHL	Y E
650	660	670 68	690	700 710	720
TCGCTGGAAGCCACG	GGCGCGGGGCAAG	Gecceeeccec	TGCTTCGGCTGCGGCGA	ACAGCAGCTACACCTACT	TCTGCGG
SLEAT	GAGK	GRAA	CFGCGD	SSYTYF	r C G
220	740	150 70	0 770	790 700	800
/ 30					000
CGCGGTGGATGCCAT	EGAAGAGCGCCT		A D T V A I	S L K I D	GDP
AVDAI		2 2 1 3			
810	820	830 84	0 850	860 870	880

0

Sall

Fig. 2. Partial nucleotide sequence of the cloned 1.6kbp fragment. The amino acid sequence deduced from the possible open reading frame is shown with one-letter abbreviations. The putative ribosomebinding site, the -35 and -10regions are indicated.

	,* ! *## **	****! *	#	#	# * *	* *# *** *
1.	MANVLIVYGS	TTGNTAWVAE	TVGRDIAEAG	HSVEIRDAGQ	VEAEGLCEGR	DLVLFGCSTW
2.	MPKALIVYGS	TTGNTEYTAE	TIARELADAG	YEVDSRDAAS	VEAGGLFEGF	DLVLLGCSTW
3.	MSKSLIVYGS	TTGNTETAAE	YVAEAFENKE	IDVELKNVTD	VSVADLGNGY	DIVLFGCSTW
4.	MSKVLIVPGS	STGNTESIAQ	KLEELIAAGG	HEVTLLNAAD	ASAENLADGY	DAVLFGCSAW
5.	MPKALIVYGS	TTGNTEGVAE	AIAKTLNSEG	METTVVNVAD	VTAPGLAEGY	DVVLLGCSTW
6.	MGKALVVFGS	TTGNTETVAE	VVAKVLEESG	MAVDLKNATK	VKAAGLAEGY	DLVVFGCSTW
7.	MSKVLILFGS	STGNTESIAQ	KLEELVAAGG	HEVTLLNAAE	ASADNLADGY	DAVLMGCSAW
	* *#* *	* *	# 1	* ** *	#**** #*##	# * *
1.	⁶¹ GDDEIELODD	FIHLYESLEA	TGAGKGRAAC	FGCGDSSYTY	FCGAVDATEE	PISCICADIV
2.	GDDSIELODD	FIPLFDSLEE	TGAOGRKVAC	FGCGDSSYEY	FCGAVDATEE	KIKNICAFIU
3.	GEEEIELODD	FIPLYDSLEN	ADLKGKKVSV	FGCGDSDYTY	FCGAVDATEE	KIEKMCAUUT
4.	GMEDLEMQDD	FLSLFEEFNR	IGLAGRKVAA	FASGDOEYEH	FCGAVPATEE	RAKELCATT
5.	GDDEIELQED	FVPLYEDLDR	AGLKDKKVGV	FGCGDSSYTY	FCGAVDVIEK	KAFFLGATIN
6.	GDDEIELQED	FIPLYDDLGA	AGLGGRKVAV	FGCGDSSYTH	FCGAVDATAE	KAASLCAKUT
7.	GMEDLELQDD	FAPLFDEMEN	MGLKGKKLAA	FASGDMEYEH	YCGAVPATEE	KAVGLGAEVI
					TOOLIVITIDD	INTO DOAL VI
	*## *		# #			
1	121 NOCT WIDODD	DOWDDDDWGDW	π π			
1.	ADSLKIDGDP	RTMRDDVSAW	AGRVVTAL			

1	ADSLKIDGDP	RTMRDDVSAW	AGRVVTAL
2.	QDGLRIDGDP	RAARDDIVGW	AHDVRGAI
3.	GDSLKIDGDP	ERDEIVGW	GSGIADKI
4.	AEGLKMEGDA	SNDPEAVASF	AEDVLKNL
5.	ASSLKIDGEP	DSAEVLDW	AREVLARV
6.	DLPLKIDGAP	DTAEARDW	AKEVLRSAA
7.	PEGLKIEGDA	SSDPDAVSAF	AEDVLK



Fig. 4. SDS-PAGE of the purified recombinant flavodoxin. Lane 1, recombinant flavodoxin; lane 2, molecular mass marker mixture.

estimated to be 0.234, so it should be proper that flavodoxin expressed in E. coli as a holoprotein binds to FMN as a prosthetic group in a molar ratio of 1.

Electrochemical and FMN-Binding Properties of the Flavodoxin and Its Mutant (A16E)-We confirmed that the recombinant flavodoxin was reduced by H2 with hydrogenase and cytochrome c_3 in vitro. The redox potentials and dissociation constants with FMN, and the riboflavin of the recombinant and mutant flavodoxin are summarized in Table I. These data indicate that these flavodoxins have basically the same characters as to prosthetic group-binding and redox potentials, and are almost the same as that from D. vulgaris (Hildenborough) (30).

Redox Behavior with a Graphite Electrode of the Wild-Type Flavodoxin-CV revealed a pair of anodic and cathodic peaks using either a BPG or EPG electrode at all the pH values, although the response was much smaller than with the flavodoxin-adsorbed EPG electrode. The formal potential of flavodoxin $(E^{\circ\prime})$ at pH 7.3 was determined to be $-477 \pm 3 \text{ mV}$ from the midpoint potential between the anodic and cathodic peak potentials as well as the peak potentials of a differential-pulse voltammogram (DPV) and



an ac voltammogram. The plot of E° as a function of pH could be fitted to three straight lines (23). The slopes were -58 mV/pH at pH < 8.5 and pH > 10, and -29 mV/pH at 8.5 < pH < 10. Therefore, the redox reaction of flavodoxin adsorbed on the EPG electrode is a two-proton transfer process at pH < 8.5 and pH > 10, while it is a two-electron one-proton transfer process at 8.5 < pH < 10. Extrapolation to pH 7.0 gave $E_{m,7}(EPG) = -435 \text{ mV}$, which is almost same as the E_1 of flavodoxin in the solution phase.

Electrochemistry of the Wild-Type Flavodoxin on a Phospholipid-Modified BPG Electrode-CV showed two pairs of anodic and cathodic peaks. The peak charge of the CV wave of the pc1-pa1 couple was approximately 1/5 that of the CV wave of the pc2-pa2 couple. From the v-dependence of the peak potentials, the $E^{\circ\prime}$ obtained for the two pairs from the CV responses were $E^{\circ'}{}_{p1} = -318 \text{ mV}$ and $E^{\circ'}{}_{p2} = -136 \text{ mV}$. Both $E^{\circ'}{}_{p1}$ and $E^{\circ'}{}_{p2}$ were more positive than $E_{m,7}(EPG)$. The plot of the peak potentials against pH in the range of 5.2 < pH < 9.2 was a straight line with a slope of -59 mV/pH, indicating that the redox reaction is a two-electron two-proton process.

Immunoblot Analysis-The recombinant flavodoxin reacted with a polyclonal antibody against the FMN-binding Downloaded from http://jb.oxfordjournals.org/ at Islamic Azad University on October 1, 2012

Fig. 3. Comparison of the amino acid sequences of flavodoxins from Desulfovibrio. Lane 1, D. vulgaris (Miyazaki F) (this work); lane 2, D. vulgaris (Hildenborough) (36, 37); lane 3, D. salexigens (12); lane 4, D. desulfuricans [Essex 6] (13); lane 5, D. gigas (ATCC19364) (14); lane 6, D. gigas (ATCC-29494) (14); lane 7, D. desulfuricans (ATCC-27774) (15). * means all identical: # means identical except for one strain; ! means identical except for D. vulgaris (Miyazaki F); . indicates a gap. The amino acid sequences of the flavodoxins from D. vulgaris (Miyazaki F) and D. vulgaris (Hildenborough) are 66.9% identical to each other.



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Fig. 6. Identification of the prosthetic group by HPLC. The extracted prosthetic group was subjected to reverse-phase HPLC on a C8 column (Hiber LiChrosorb RP-8) with 10% acetonitrile in 0.1% trichloroacetic acid at the flow rate of 2 ml/min.

TABLE I. The dissociation constants of apoprotein with FMN and riboflavin, and the redox potentials of the recombinant and mutant (A16E) flavodoxins. K_d stands for the dissociation constant, and E_1 and E_2 stand for the redox potentials between hydroquinone and semiquinone, and semiquinone and the fully oxidized state, respectively.

		Wild type	A16E	
K_{d} with FMN	(nM)	0.38	0.36	
K_{d} with riboflavin	(mM)	2.13	2.16	
E_1	(mV)	-434	-425	
E_2	(mV)	-151	-136	
				_

protein 100 times less than the recombinant FMN-binding protein (see Fig. 7). This suggests that the higher structures of the flavodoxin and FMN-binding protein are not similar enough for the FMN-binding protein antibody to crossreact.

DISCUSSION

In this study, we isolated the gene encoding a flavodoxin from D. vulgaris (Miyazaki F) by means of PCR. This is the first report of determination of the amino acid sequence of the flavodoxin from D. vulgaris (Miyazaki F). Our results indicate that the sulfate-reducing bacteria commonly contain flavodoxin with highly homologous amino acid sequences, and that they have some conserved sequences around the FMN-binding region. However, the redox and FMN-binding properties of the A16E mutant indicated that the conserved Glu-16 residue is not particularly important, although it is located near the FMN-binding region and is on the surface of the molecule. The A16E mutant may interact with hydrogenase in a different manner from the wild type. Recently, the electrostatic effects of surface Glu or Asp residues on redox potentials of the flavodoxin from D. vulgaris (Hildenborough) were reported, and the redox potential between semiguinone and hydroquinone was correlated with the number of acid-to-amide substitutions (32). However, the redox properties (both E_1 and E_2) of the



Fig. 7. Immunoblot analysis of the recombinant flavodoxin and FMN-binding protein. Lane 1, purified FMN-binding protein, 8 pmol; lane 2, purified FMN-binding protein, 4 pmol; lane 3, purified FMN-binding protein, 0.8 pmol; lane 4, purified flavodoxin, 800 pmol; lane 5, purified flavodoxin, 400 pmol; lane 6, purified flavodoxin, 80 pmol.

A16E mutant appear to be similar to those of the wild-type flavodoxin. The relationship between the number of acidic amino acid residues and the redox properties may not be so simple.

It is known that free FMN adsorbed on an EPG electrode exhibits two pairs of anodic and cathodic peaks under the same conditions (Sagara, T. and Takaki, S., unpublished data). Therefore, the observed voltammetric response with a bare graphite electrode could be attributable to the redox response of FMN bound to the peptide, *i.e.* flavodoxin. Taken together with the appearance of the single peak in the voltammograms, the semiguinone state of the flavodoxin adsorbed on a bare EPG electrode appears to be unstable, as opposed to the flavodoxin in the solution phase. In the case of free FMN, CV with a phospholipid-modified BPG electrode gives rise to a sigmoidal-shaped wave, even at a low v (Sagara, T., Takaki, S., Honda, M., Ezoe, K., and Nakashima, N., unpublished data). Because the CV wave of flavodoxin on the phospholipid-modified BPG electrode at a low v was peak-shaped, the CV response obtained was not due to free FMN released from flavodoxin. Although two pairs of CV responses were observed, it would not be correct to equate $E^{\circ'}{}_{p1}$ and $E^{\circ'}{}_{p2}$ to E_1 and E_2 , respectively. The appearance of the two pairs may rather indicate that two states of flavodoxin coexisted in the phospholipid film. It is note worthy that the redox activity of the flavodoxin adsorbed on the EPG electrode was stable for a week, although a slight decrease in the peak current over time was observed. Therefore, it may be promising for electroengineering as a bio-tip when recombinant DNA techniques are being used.

All flavodoxins have a typical structure, that is, a fivestranded parallel β -sheet flanked on either side by a pair of α -helices. FMN lies mostly below the surface of the molecule, and the isoalloxazine ring of FMN appears to be planar, and is buried between two segments of the polypeptide chain and the side groups, Trp-60 and Tyr-98 (7). Tyr-98 is especially important in modulating the redox potential by means of site-directed mutagenesis (33). These two amino acids are also conserved in the flavodoxin from D. vulgaris (Miyazaki F), which is thought to have a structure similar to those of other flavodoxins. However, immunoblot analysis clearly indicated that flavodoxin is a different molecule from the FMN-binding protein in the higher structure, although they both have a similar amino acid sequence region (Thr-Trp-Asn or Gly). Gly-61 of flavodoxin has been reported to be important in stabilizing the semiquinone state (34), while the Trp residue in this region is important for FMN-binding in the case of the FMN-binding protein (K_d with FMN of the wild-type FMN-binding protein is 0.43 nM, unpublished data). We are now studying the structures of the FMN-binding protein and flavodoxin in detail by NMR, and preliminary data have indicated that they are not similar to each other in the interaction between FMN and the peptide chain (35). Although both the FMN-binding protein and flavodoxin have almost the same molecular weight and bind to FMN with almost the same dissociation constant from the same strain of D. vulgaris, the FMN-binding protein is a different molecule from flavodoxin based on the amino acid sequence, higher structure, and redox properties.

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